225. Synthesis and Biological Properties of N-Acetyl-4-deoxy-d-neuraminic Acid

by Heinz-Werner Hagedorn and Reinhard Brossmer*

Institut für Biochemie II, Universität Heidelberg, Im Neuenheimer Feld 328, D-6900 Heidelberg

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N-Acetylneuraminic acid (1) can be transformed into the methyl α -D-ketoside 2 which, by reaction with methanesulfonyl chloride, yields the corresponding 4-*O*-mesylate 3 and the 4,7-di-*O*-mesylate 4 as a by-product. Compound 3 reacts with NaI giving the 4-deoxy-4-iodo compound 5 with equatorial orientation of the I-atom. As second product, the dihydrooxazole 6 is produced. Catalytic hydrogenation of 5 is followed by ester cleavage and removal of the isopropylidene group yielding the methyl α -D-ketoside 8 which affords the title compound, *N*-acetyl-4-deoxyneuraminic acid (9), by reaction with *fowl plague virus* sialidase. Further biochemical activities of 8 and 9 are reported.

Introduction. – Sialic acids as constituents of glycoproteins and glycolipids are of considerable importance for a multitude of biological phenomena: cell-cell recognition, cell adhesion, interactions within the immune system and between virus and host cell, to mention only a few examples. For some years, this laboratory is carrying out a program aiming at the relationship between chemical structure and biological activity of sialic acids. This work has resulted in the synthesis of sialic-acid analogues which are modified at the particular regions of the parent molecule.

Position C(4) in *N*-acetylneuraminic acid (Neu5Ac; 1) appears to be particularly sensitive to the action of some enzymes involved in the metabolism of 1. Thus, α -keto-sides of *N*,4-*O*-diacetylneuraminic acid (Neu4,5Ac₂) are not attacked by bacterial sialidases, but are slowly cleaved by viral sialidases [1][2]. Ketosides of *N*-acetyl-4-*O*-methylneuraminic acid (Neu4Me5Ac) are completely resistent towards bacterial sialidases, but are cleaved rapidly by viral sialidases [1]. Neu4Me5Ac was activated to the corresponding CMP-glycoside and transferred onto asialo glycoprotein [1]. In contrast, Neu4,5Ac₂ apparently was not activated [3]. Neu4,5Ac₂ is poorly cleaved, and Neu4Me5Ac is not cleaved at all by the action of aldolase [1]. The biological properties of some further C(4) analogues have been reported [4–6]. In our laboratory, synthesis and biological properties of two additional substrates, 5-acetamido-3,5-dideoxy-D-glycero-D-talo-2-nonulosonic acid (4-epi-Neu5Ac) [7] and 5-acetamido-3,5-dideoxy-D-manno-2-non-4-ulosonic acid (4-oxo-Neu5Ac) [8] have been studied.

Here, we report on the synthesis of novel O-devoided C(4) analogues of 1 [9] by some appropriate steps starting with the natural compound itself.

Results and Discussion. – A simple and approved standard reaction for modifying OH groups is to prepare sulfonate esters which are prone to nucleophilic substitution at the same C-atom. The more it is surprising that this common method never has been used to modify the substitution pattern at C(4) of *N*-acetylneuraminic acid (1). To achieve this, 1

had to be protected in a convenient way. According to the method of *Kuhn et al.* [10], **1** was transformed into the per-*O*-acetylated methyl β -D-ester followed by replacement of the anomeric acetyl group with a Cl-atom. Reaction with MeOH catalyzed by polymeric silver salt gave the corresponding methyl α -D-ketoside [11] which was de-*O*-acetylated by *Zemplén* saponification affording methyl *N*-acetyl-2-*O*-methyl- α -D-neuraminate [12] [13]. Finally, OH-C(8) and OH-C(9) were protected by an isopropylidene group to give **2** [14] (*Scheme 1*). This made sure that only OH-C(7) and OH-C(4) were left for the following reactions.

According to our experience, OH-C(7) of derivatives like 2 proves to be less reactive than OH-C(4), whereupon a selective transformation at C(4) should be possible. In fact, mesylation of 2 yielded the crystalline 4-O-mesylate 3 (50%), and the 4,7-di-O-mesylate 4 was found as a by-product (15%), as expected (Scheme 1).



According to Yamasaki et al. [15], methyl 2-acetamido-4,6-di-O-acetyl-2-deoxy-3-O-tosyl- β -L-glucopyranoside readily reacts with NaI in 74% yield to give the 3-deoxy-3-iodo derivative as the only product. Nevertheless, the authors did not report any spectroscopic data in detail. Hence, the configuration at C(3) of this compound remains uncertain.

A comparable *Finkelstein* reaction could be applied to the mesylate 3. However, the corresponding 4-iodo compound 5 was isolated, after chromatographic separation and recrystallization from CH₂Cl₂/light petroleum, only in 20% yield (*Scheme 2*). From *Beilstein*'s test and combustion analysis, the presence of an I-atom was deduced. Surprisingly, the I-atom of 5 has exclusively equatorial and not axial orientation at C(4) as shown by the ¹H-NMR spectrum (J(3a,4) = J(4,5) = 12.8 Hz). In comparison with 3, H–C(4) is shifted clearly upfield ($\Delta \delta = 0.66$ ppm). Furthermore, the structure of 5 was confirmed by X-ray analysis [16]. With these results, it became clear that the substitution does not follow a simple S_82 mechanism accompanied by *Walden* inversion. This assumption was further substantiated by detection of dihydrooxazole 6 which was isolated as a crystalline compound (25%), after chromatographic separation.

The ¹H-NMR spectrum of **6** shows at 1.97 ppm the *d* of the CH₃ group at the oxazole moiety with ${}^{5}J(5,CH_{3}) = 0.9$ Hz. Other, noteworthy coupling constants are J(3a,4) = 9.7 and J(3e,4) = 6.6 Hz. All results are in accordance with the postulated structure of **6**.

Brandstetter and Zbiral [17] prepared the methyl β -D-ketoside analogous to **6** in a different way. But it could not be characterized sufficiently because of lacking stability. Only when OH-C(7) was acetylated, this dihydrooxazole became stable and could be analyzed by ¹H-NMR spectroscopy.



Obviously, mesylate **3**, before reacting with NaI, is attacked at C(4) by the *N*-acetyl group (neighbouring-group participation) which represents an intramolecular $S_{\aleph}2$ mechanism (Scheme 2). The resulting dihydrooxazolium ion **3a** is then converted into the dihydrooxazole **6** by loss of a proton or, alternatively, is attacked at C(4) by I⁻ thereby regenerating the acetylamino group at C(5) (Scheme 2).

Catalytic hydrogenation of the iodo compound 5 with H_2/Pd readily gave the crystalline 4-deoxy compound 7 in a yield of 84% (*Scheme 3*). The structure of 7 has been confirmed by ¹H-NMR spectroscopy (H_{ax} -C(4) (1.45 ppm) and H_{eq} -C(4) (1.96 ppm) appear as *dddd*'s with J(4ax, 4eq) = 13.3 Hz).



Subsequently, the methyl ester at C(1) and the 8,9-O-isopropylidene group of 7 were cleaved in a one-pot reaction. First, 7 was treated with 1N NaOH, and after complete saponification of the ester, an excess of *Dowex 50 WX 4* (H⁺) was added. Finally, methyl ketoside 8 was purified by ion-exchange chromatography on *DEAE Sephadex A-25* (HCO₃⁻). The results of the deblocking reactions have been proven by the ¹H-NMR spectrum of 8 (the characteristic s of the isopropylidene group and of the ester of 7 are absent).

In order to obtain free *N*-acetyl-4-deoxyneuraminic acid (9), the ketoside had to be cleaved. Although there are approved chemical procedures [18], the particularly careful enzymatic cleavage by a suitable sialidase was recommended as 9 is rather sensitive to acid. At the same time, the biological activity of 8 could be studied. The reaction of 8 with fowl plaque virus sialidase proved to be promising and was complete after 20 h at 37°. After purification by ion-exchange chromatography on *DEAE Sephadex A-25* (HCO₃⁻) and freeze-drying, 9 was isolated in a yield of 92%.

In the ¹H-NMR spectrum of 9, CH₂(3), CH₂(4), and CH₃CONH are found as a *m* at 2.09–1.95 ppm which could not be analyzed. H–C(5) and H–C(6) give also a *m* at 4.12–4.07 ppm, whereas the signals of the protons at C(7), C(8), and C(9) are coming up clearly.

We found that ketoside 8 is not a substrate for bacterial sialidases from *Vibrio* cholerae, Clostridium perfringens, and Arthrobacter ureafaciens. However, 8 was also cleaved by sialidase from bovine testis [19].

Furthermore, the biological activity of **9** was studied. As expected, **9** was not cleaved by *N*-acetylneuraminate pyruvate-lyase (EC 4.1.3.3), but proved to be an inhibitor for the cleavage of the natural compound **1** (inhibition rate 80%, **1/9** ratio 1:2) [19].

Interestingly, **9** was activated to the corresponding CMP-glycoside by CMP-sialicacid synthase and was transferred onto asialo- α_1 -acid glycoprotein by α -2,6-sialyltransferase from rat liver [20].

In this study, we were able to prepare novel C(4) analogues of *N*-acetylneuraminic acid which show interesting properties and thus require further investigations of the biological relevance of the C(4) OH group.

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Experimental Part

General. All chemicals were obtained from commercial sources. Fowl plague virus sialidase was a generous gift from Prof. R. Rott, Universität Giessen. Solvents were distilled before use. N-Acetylneuraminic acid (1) was prepared from edible-bird's nests [21]. Solns. were evaporated in a rotary evaporator below 40°. TLC: Merck precoated plates, silica gel 60 F_{254} , layer thickness 0.2 mm; detection by spraying with 2N H₂SO₄ followed by heating at 150°. Column chromatography: Merck silica gel 60, 0.063-0.200 mm. M.p.: Tottoli-Büchi apparatus; uncorrected. Optical rotations: Perkin-Elmer-141 polarimeter. ¹H-NMR spectra: if no mention, on Bruker WM-300 (¹H(300 MHz)), otherwise on Bruker AM-500 (¹H(500 MHz)) using tetramethylsilane (TMS) as the internal standard; δ in ppm relative to TMS; coupling constants were determined first order. Elemental analysis was performed in the Max-Planck-Institut für Medizinische Forschung, Heidelberg.

Methyl (Methyl 5-Acetamido-3,5-dideoxy-8,9,O-isopropylidene- α -D-glycero-D-galacto-2-nonulopyranosid)onate (2). To a soln. of methyl (methyl 5-acetamido-3,5-dideoxy- α -D-glycero-D-galacto-2-nonulopyranosid)onate [12] (6.0 g, 17.8 mmol) in anh. acetone (100 ml), TsOH · H₂O (100 mg, 0.53 mmol) was added and stirred at r.t. for 4 h. Then, *Dowex 1* × 4 (AcO⁻, 1.0 g) was added, and vigorous stirring was continued for 15 min. Finally, the resin was filtered off and washed with acetone (3 × 10 ml). After removing of the solvent, the residue was crystallized from MeOH/Et₂O affording 5.0 g (75%) of **2**, m.p. 181°, R_f 0.55 (CHCl₃/MeOH 4:1), $[\alpha]_D^{20} = -9.0^\circ$ (c = 0.5, CHCl₃). ¹H-NMR (500 MHz, CDCl₃/D₂O): 4.30 (*ddd*, J = 6.3, 6.3, 6.3, H-C(8)); 4.12 (*dd*, J = 8.6, 6.3, H-C(9)); 3.85 (*m*, H–C(5)); 3.81 (*s*, CO₂CH₃); 3.72 (*m*, H–C(4)); 3.57 (*dd*, J = 6.3, 1.3, H–C(7)); 3.51 (*dd*, J = 10.4, 1.3, H-C(6)); 3.39 (*s*, CH₃O); 2.69 (*dd*, J = 12.9, 4.8, H-C(3)); 2.06 (*s*, AcN); 1.83 (*dd*, J = 12.9, 11.0, H-C(3)); 1.40, 1.37 (2*s*, (CH₃)₂C). Anal. calc. for C₁₆H₂₇NO₉ (377.39): C 50.92, H 7.21, N 3.71; found: C 50.99, H 7.24, N 3.61.

Methyl (Methyl 5-Acetamido-3,5-dideoxy-8,9-O-iosopropylidene-4-O-methanesulfonyl- α -D-glycero-D-galacto-2-nonulopyranosid) onate (3) and Methyl (Methyl 5-Acetamido-3,5-dideoxy-8,9-O-isopropylidene-4,7-di-O-methanesulfonyl- α -D-glycero-D-galacto-2-nonulopyranosid) onate (4). To a soln. of **2** (950 mg, 2.52 mmol) in anh. pyridine (10 ml) cooled in ice, methanesulfonyl chloride (200 µl, 2.58 mmol) was added dropwise during 40 min and the mixture stored in the refrigerator at -10° overnight. The soln. was poured on ice/H₂O (20 ml), extracted with CH₂Cl₂ (3 × 20 ml), dried (MgSO₄), and evaporated. The residue was co-distilled with toluene (10 ml) and gave, after chromatography (toluene/AcOEt/EtOH 5:1:1), 580 mg (51%) of 3 (recrystallized from AcOEt/light petro-leum) and 200 mg (15%) of **4**. *Data of* **3**: M.p. 133–134°, $R_{\rm f}$ 0.33, [α]_D²⁰ = -19.2° (c = 1.0, CHCl₃). ¹H-NMR (CDCl₃/D₂O): 4.87 (ddd, J = 7.5, 6.2, 5.7, H–C(8)); 4.86 (dd, J = 11.5, 11.5, 5.3, H–C(4)); 4.12 (dd, J = 8.4, 6.2, H–C(9)); 4.05 (dd, J = 8.4, 5.7, H–C(9)); 4.01 (dd, J = 11.5, 10.4, H–C(5)); 3.84 (s, CO₂CH₃); 3.52 (d, J = 7.5, H–C(7)); 3.45 (s, CH₃O₁); 3.39 (d, J = 10.4, H–C(6)); 3.10 (s, MsO); 2.78 (dd, J = 12.4, 5.3, H–C(3)); 2.06 (s, AcN); 1.38, 1.36 (2s, (CH₃)₂C). Anal. calc. for C₁₇H₂₉NO₁₁S (455.48): C 44.83, H 6.42, N 3.07; found: C 45.00, H 6.42, N 3.01.

Data of 4: $R_{\rm f} 0.41$, $[\alpha]_{\rm D}^{20} = +5.25^{\circ}$ (c = 0.5, CHCl₃). ¹H-NMR (CDCl₃/D₂O): 5.45 (ddd, J = 11.5, 9.7, 5.3, H–C(4)); 5.11 (dd, J = 2.4, 1.1, H–C(7)); 4.62 (dd, J = 10.6, 1.1, H–C(6)); 4.38 (ddd, J = 8.0, 6.6, 2.4, H–C(8)); 4.18 (dd, J = 8.4, 6.6, H–C(9)); 4.04 (dd, J = 8.4, 8.0, H–C(9)); 3.88 (s, CO₂CH₃); 3.34 (s, MsO); 3.30 (dd, J = 10.6, 9.7, H–C(5)); 3.22 (s, MsO); 3.01 (s, CH₃O); 2.92 (dd, J = 12.8, 5.3, H–C(3)); 1.99 (s, AcN); 1.90 (dd, J = 12.8, 11.5, H–C(3)); 1.44, 1.37 (2s, (CH₃)₂C). Anal. calc. for C₁₈H₃₁NO₁₃S₂ (533.57): C 40.52, H 5.86, N 2.62; found: C 40.29, H 5.58, N 2.67.

Methyl (Methyl 5-Acetamido-3,4,5-trideoxy-4-iodo-8,9-O-isopropylidene- α -D-glycero-D-galacto-2-nonulopyranosid) onate (5) and Methyl (Methyl 3,4,5-Trideoxy-4,5-dihydro-8,9-O-isopropylidene-2'-methyloxazolo[5,4-d]- α -D-glycero-D-talo-2-nonulopyranosid) onate (6). A soln. of 3 (500 mg, 1.1 mmol) and anh. Na1 (5.00 g, 39.4 mmol) in anh. acetone (40 ml) was heated in a steel bomb for 3 h at 100°. After cooling, the suspension was evaporated, and the solid residue was extracted with a mixture of sat. sodium thiosulfate soln. (20 ml) and CH₂Cl₂ (2 × 20 ml). The combined org. layers were dried (MgSO₄) and evaporated. The yellow-brown syrup left yielded, after chromatography (AcOEt), 105 mg (20%) of 5 (recrystallized from CH₂Cl₂/light petroleum) and 100 mg (25%) of 6 (recrystallized from AcOEt/light petroleum). Data of 5: M.p. 140–141°, R_f 0.43, $[\alpha]_D^{20} = +18.0°$ (c = 0.5, CHCl₃). ¹H-NMR (CDCl₃/D₂O): 4.27 (ddd, J = 12.4, 6.2, 6.2, H–C(8)); 4.20 (ddd, J = 12.8, 12.8, 4.4, H–C(4)); 4.14–4.05 (m, 2 H–C(9)); 4.05 (dd, J = 12.8, 10.2, H–C(5)); 3.82 (s, CO_2CH_3); 3.60 (d, J = 6.2, H–C(7)); 3.52 (d, J = 10.2, H–C(6)); 3.40 (s, CH_3O); 3.08 (dd, J = 13.4, 4.4, H–C(3)); 2.53 (dd, J = 13.4, 12.8, H–C(3)); 2.09 (s, AcN; 1.39, 1.36 (2 $s, (CH_3)_2C$). Anal. calc. for C₁₆H₂₆INO₈ (487.29): C 39.44, H 5.38, I 26.04, N 2.87; found: C 39.21, H 5.26, I 26.11, N 2.83.

Data of 6: M.p. 115°, $R_f 0.15$, $[\alpha]_{20}^{20} = -5.3°$ (c = 0.5, CHCl₃). ¹H-NMR (CD₃OD): 4.88 (m, H–C(4)); 4.46 (m, H–C(5)); 4.27 (ddd, J = 6.6, 6.2, 6.2, H–C(8)); 4.12 (dd, J = 8.4, 6.2, H–C(9)); 4.00 (dd, J = 8.4, 6.6, H–C(9)); 3.85 (dd, J = 6.2, 1.1, H–C(7)); 3.77 (s, CO₂CH₃); 3.68 (dd, J = 10.2, 1.1, H–C(6)); 3.35 (s, CH₃O); 2.51 (dd, J = 13.9, 6.6, H–C(3)); 2.12 (dd, J = 13.9, 9.7, H–C(3)); 1.97 (d, J = 0.9, CH₃–C(2')); 1.38, 1.34 (2s, (CH₃)₂C). Anal. calc. for C₁₆H₂₄NO₈ (358.37): C 53.62, H 6.75, N 3.91; found: C 53.50, H 7.02, N 3.82.

Methyl (Methyl 5-Acetamido-3,4,5-trideoxy-8,9-O-isopropylidene- α -D-manno-2-nonulopyranosid) onate (7). A soln. of 5 (50 mg, 0.1 mmol) in anh. MeOH (8 ml) and anh. pyridine (20 µl) was hydrogenated over H₂/Pd during 2 d at r.t. After evaporation of the pale yellow soln., the residue was dissolved in CHCl₃ (15 ml) and shaken with a soln. of sat. sodium thiosulfate (8 ml). The aq. layer was re-extracted with CHCl₃ (10 ml). The combined org. layers were dried (MgSO₄) and evaporated affording a syrup which slowly crystallized. Recrystallization from Et₂O/light petroleum yielded 31 mg (84%) of 7, m.p. 120–121°, R_{f} 0.47 (AcOEt), $[\alpha]_{D}^{20} = +2.8^{\circ}$ (c = 0.5, CHCl₃). ¹H-NMR (CD₃OD): 4.27 (m, H–C(8)); 4.08 (dd, J = 8.4, 62, H–C(9)); 4.00 (dd, J = 8.4, 66, H–C(9)); 3.95 (ddd, J = 11.5, 11.5, 4.0, H–C(5)); 3.80 (s, CO₂CH₃); 3.64–3.60 (m, H–C(6), H–C(7)); 3.33 (s, CH₃O); 2.30 (ddd, J = 13.7, 4.0, 4.0, 4.0, 4.0, H–C(4)); 1.94 (s, AcN); 1.72 (ddd, J = 13.7, 13.3, 4.0, H–C(3)); 1.45 (dddd, J = 13.3, 13.3, 11.5, 4.0, H–C(4)); 1.37, 1.34 (2s, (CH₃)₂C). Anal. calc. for C₁₆H₂₇NO₈ (361.39): C 53.18, H 7.53, N 3.88; found: C 53.03, H 7.72, N 3.96.

Methyl 5-Acetamido-3,4,5-trideoxy- α -D-manno-2-nonulopyranosidonic Acid (8). To a stirred soln. of 7 (310 mg, 0.86 mmol) in MeOH (1 ml) cooled in ice, 1N NaOH (1.5 ml) was added dropwise and then, the stirring was continued for 2 h at r.t. After addition of H₂O (2 ml), the soln. was acidified with *Dowex WX* 4 (H⁺) to pH 2.5 and vigorously stirred for 2.5 h. The resin was filtered off, intensively washed with H₂O (30 ml), and the combined filtrate was freeze-dried. For purification, the solid residue was transferred onto a column filled with *DEAE* Sephadex A-25 (HCO₃⁻), washed with H₂O (70 ml), and eluted with 0.075M NH₄HCO₃. The fractions containing 8 were collected and evaporated (bath temp. max. 20°). A soln. of the residue in H₂O (20 ml) was rapidly acidified with *I* 20 (H⁺) to pH 2.5, then filtered, and the resin was washed intensively with H₂O (30 ml). The combined filtrates were freeze-dried giving 200 mg (72%) of 8, R_f 0.55 (EtOH/ACOH 5:1), $[\alpha]_D^{20} = -7.8^\circ$ (c = 0.5, H₂O). ¹H-NMR (500 MHz, D₂O): 3.93–3.87 (m, H–C(5), H–C(8)); 3.87 (dd, J = 11.8, 2.6, H–C(9)); 3.81 (dd, J = 10.4, 1.7, H–C(6)); 3.64 (dd, J = 11.8, 6.3, H–C(9)); 3.57 (dd, J = 8.9, 1.7, H–C(7)); 3.35 (s, CH₃O); 2.36 (ddd, J = 13.5, 13.5, 12.0, 4.3, 3.7, H–C(4)); 1.98 (s, AcN); 1.71 (ddd, J = 13.5, 13.5, 12.0, 3.7, H–C(4)). Anal. calc. for C₁₂H₂₁NO₈ · 1 H₂O (325.31): C 44.31, H 7.13, N 4.31; found: C 44.78, H 7.17, N 4.43.

5-Acetamido-3,4,5-trideoxy-D-manno-2-nonulosonic Acid (9). Acid 8 (36.1 mg, 0.12 mmol) was dissolved in 0.1M NaOAc buffer (1.8 ml), and the pH was focused to 6.0 by addition of 1N NaOH. After addition of fowl plague virus sialidase (0.5 U), the soln. was incubated 20 h at 37°. When TLC (EtOH/AcOH 5:1) indicated complete cleavage of 8, the soln. was dialyzed 6 times each 0.5 h vs. H₂O (100 ml). The dialysates were transferred onto a column filled with *DEAE Sephadex A-25* (HCO₃⁻), washed with H₂O (100 ml), and eluted with 0.08M NH₄HCO₃ soln. The fractions containing 9 were collected, evaporated (bath temp. max. 25°) and dissolved in H₂O (15 ml). The soln. was acidified with *IR 120* (H⁺) to pH 2.5 and freeze-dried yielding 31.8 mg (92%) of 9, R_f 0.26 (EtOH/AcOH 5:1), $[\alpha]_{D}^{20} = -38.0^\circ$ (c = 0.4, H₂O, $D_{K_a} = 3.30$ (H₂O, 22°). ¹H-NMR (D₂O): 4.12-4.07 (m, H-C(5), H-C(6)); 3.92 (dd, J = 11.5, 2.6, H-C(9)); 3.85 (ddd, J = 8.8, 6.2, 2.6, H-C(8)); 3.70 (dd, J = 11.5, 6.2, H-C(9)); 3.66 (dd, J = 8.8, H-C(7)); 2.09-1.95 (m, 2 H-C(3), 2 H-C(4), AcN). Anal. calc. for C₁₁H₁₉NO₈·1 H₂O (311.29): C 42.44, H 6.80, N 4.50; found: C 42.20, H 6.46, N 4.57.

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